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Inhibition of the strand transfer step of HIV-1 integrase by non-natural dinucleotides

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Abstract—New, non-natural dinucleotide 5'-monophosphates, with a surrogate isonucleoside component of L-related stereochemistry at the 'terminal' position, have been synthesized. Structures of 2a–c were confirmed by multinuclear NMR spectra (¹H, ¹³C, ³¹P, COSY), UV hypochromicity and FAB HRMS data. These compounds are totally resistant to cleavage by 3'- and 5'-exonucleases. The dinucleotides showed remarkable selectivity for inhibition of the strand transfer step of HIV-1 integrase. To the best of our knowledge, these compounds represent only the second example of selective strand transfer inhibitors of HIV integrase.

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The retroviral enzyme, HIV-1 integrase, incorporates HIV double helical DNA into host chromosomal DNA. 1-7 This viral enzyme first catalyzes the endonucleatic removal of two terminal nucleotides at the 3'-end of each strand of viral DNA (3'-processing) leaving recessed ends that terminate with xxCA-OH (Fig. 1). In the next steps (strand transfer, integration) nucleophilic attack of the terminal 3'-OH of the tailored HIV DNA on a specific internucleotide phosphodiester functionality results in cleavage of host DNA and this is followed by integration of the tailored HIV DNA into host DNA. 1-3 The integration process is essential for the replication of HIV and there is apparently no functional equivalent of HIV integrase in human cells.

Endonuclease Activity

↓

LTR (+) 5'-end ACTG......CAGT (+) 3'-end

(-) 3'-end TGAC......GTCA (-) 5'-end

↑

Endonuclease Activity

Figure 1. Processing of HIV DNA prior to integration into host DNA.

Some oligonucleotides of natural origin are capable of interfering with the integration process by competing with viral DNA for binding to HIV integrase. Protein–nucleotide interactions appear to be of importance in other steps of the replication cycle of HIV such as the recognition and binding of Tat protein to HIV-1 TAR RNA. However, small oligonucleotides of natural origin are rapidly cleaved by cellular nuclease activity. In addition, increasing nuclease resistance by chemical alteration of the internucleotide phosphate bond results in decreased integrase activity. A non-natural dinucleotide with a conformationally unusual internucleotide phosphodiester bond that joins a D-deoxynucleoside and an L-related isodeoxynucleoside, 1 (Fig. 2).

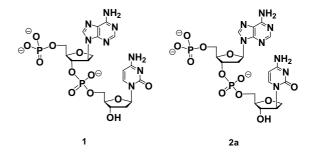


Figure 2. Structure of the anti-HIV integrase compound 1 and its isomer 2a in which the position of the surrogate isonucleoside is changed.

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Scheme 1. Reagents and conditions: [i] (*i*-Pr₂N)₂POCH₂CH₂CN, 1H-tetrazole, CH₂Cl₂, rt, 1h; [ii] (1) 1H-tetrazole, 3h, (2) I₂/THF/H₂O/Py; (3) 2% CHCl₂COOH in CH₂Cl₂; [iii] (1) *i*-Pr₂NP(OCH₂CH₂CN)₂, 1H-tetrazole, CH₂Cl₂, 1h; (2) I₂/THF/H₂O/Py; [iv] concd NH₄OH, rt, 24h.

previously synthesized by us, ¹⁰ exhibits resistance to mammalian 3'- and 5'-exonucleases. ¹¹ This compound is an inhibitor of wild-type HIV-1 integrase activity, inhibiting both the 3'-processing and strand transfer steps. ¹⁰ We report here some new results that suggest that changing the position of the surrogate nucleoside component of the dinucleotide (e.g., compound 2a) can dramatically change the mode of inhibitory activity from both key steps of integrase action to just the strand transfer step.

The isonucleosides were prepared from ribitol cyclic sulfate as the precursor 12 and then were appropriately tailored for the coupling steps. Dinucleotides were synthesized through the phosphoramidite method Scheme 1).10 Thus, isodeoxynucleoside 3 with a free 5'-hydroxyl group was condensed with the phosphorylating reagent [(i-Pr₂N)₂POCH₂CH₂CN] in the presence of 1H-tetrazole to give the phosphoramidite 4, which was immediately coupled through the free 3'-hydroxyl group of deoxynucleoside 5. Iodine oxidation and selective removal of the trityl group using dichloroacetic acid¹⁰ provided the phosphotriester **6**. Phosphorylation of dinucleotide 6 was performed using di(2-cyanoethyl)N,N-diisopropyl-phosphoramidite as the phosphorylating reagent.¹³ After deprotection, the final products were purified by HPLC. The structures of **2a**-c were confirmed by multinuclear NMR data (¹H, ¹³C, ³¹P, COSY) and FAB HRMS data. ¹⁴ The observed hypochromicity (12-18%) from the quantitative UV data [2a, λ_{max} 262 (ϵ 19,900); 2b, λ_{max} 259 (ϵ 26,400); **2c**, λ_{max} 259 (ε 24,500)] suggested the presence of base

Table 1. Anti-HIV-1 integrase data for dinucleotides

Compounds	3'-Processing, IC ₅₀ (μM)	Strand transfer, IC_{50} (μM)
2a	>1000	65
2b	>1000	41
2c	>1000	114
1	19	25

stacking in the preferred conformations. Base stacking implies the presence of conformationally unusual internucleotide phosphate bonds because of the spatial arrangement of the two sugar rings that is necessary to accommodate this assembly of the bases (Fig. 2).

Integrase inhibition assays were conducted with purified recombinant HIV-1 integrase using a 21-mer oligonucleotide substrate. 15 The data (Table 1) clearly showed that compounds 2a-c have strand transfer inhibitory activity against wild-type HIV-1 integrase but do not exhibit inhibition of the 3'-processing step. This is in sharp contrast to compound 1 (pIsodApdC), which showed strong inhibition of both key steps of the integrase mechanism of action. The major structural difference between dinucleotide 1 and its counterpart 2a (or the related compound 2b) is the position of the surrogate isonucleoside component. The inhibition of integrase by 1 is likely the result of base recognition and binding by the viral enzyme. Thus, it is remarkable that this apparently small structural change in the counterpart of 1 can produce such a major impact on the mode of inhibition of integrase. The only reported selective inhibitor of strand transfer is a class of diketo containing compounds, 16 and, to the best of our knowledge, the compounds described in this communication represent the second examples. Finally, compounds 2a-c are resistant to cleavage of the internucleotide phosphodiester linkage by mammalian 3'- and 5'-exonucleases.

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- 14. Experimental data: **2a**, ¹H NMR (D₂O): 8.40 (s, 1H), 8.20 (s, 1H), 7.74 (d, 1H, J = 7.5), 6.37 (t, 1H, J = 6.5), 6.04 (d, 1H, J = 7.5), 4.82 (m, 1H), 4.73 (m, 1H), 4.28 (m, 2H), 3.98-4.09 (m, 3H), 3.89-3.93 (m, 3H), 3.81 (m, 1H), 2.71 (m, 1H), 2.61 (m, 1H). ¹³C NMR (D₂O): 159.2, 150.5, 149.2, 148.3, 146.4, 145.5, 142.4, 118.5, 95.2, 85.5, 84.7, 84.3, 75.98, 75.92, 69.0, 65.1, 64.8, 64.1, 38.8. ³¹P NMR: 0.94, -0.030. FAB-HRMS: $[M+1]^+$ calcd for $C_{19}H_{27}N_8O_{12}P_2$ 621.1224, found 621.1217. UV (H₂O): λ_{max} 262 (ϵ 19,900). **2b**, ¹H NMR (D₂O): 8.37 (s, 1H), 8.26 (s, 1H), 8.19 (s, 1H),
 - 8.16 (s, 1H), 6.29 (t, 1H, J = 7.0), 4.98 (m, 1H), 4.80 (s, br,

- 1H), 4.47 (m, 1H), 4.21-4.28 (m, 3H), 3.90-4.04 (m, 5H), 2.63–2.66 (m, 1H), 2.55–2.58 (m, 1H). ¹³C NMR (D₂O): 150.2, 150.1, 148.5, 148.0, 145.1, 144.8, 142.8, 142.4, 118.4, 118.1, 85.3, 84.6, 84.0, 76.1, 75.8, 69.8, 64.6, 64.1, 62.8, 38.7. ³¹P NMR: 0.065, -0.75. FAB-HRMS: [M+1]⁺ calcd for C₂₀H₂₇N₁₀O₁₁P₂ 645.1336, found 645.1325. UV (H₂O): λ_{max} 259 (ϵ 26,400).
- 2c, H NMR (D₂O): 8.39 (s, 1H), 8.34 (s, 1H), 8.19 (s, 1H), 8.16 (s, 1H), 6.30 (t, 1H, J = 6.5), 5.21 (m, br, 1H), 4.82 (s, br, 1H), 4.27 (s, br, 1H), 4.21 (s, br, 1H), 4.14 (d, 1H, J = 10.5), 4.04 (m, 1H), 3.98 (dd, 1H, J = 10.5, 5.5), 3.86– 3.92 (m, 3H), 2.67 (m, 2H), 2.58 (m, 1H), 2.11 (m, 1H). ¹³C NMR (D₂O): 150.4, 150.1, 148.4, 148.2, 145.3, 144.8, 142.9, 142.4, 118.5, 118.1, 85.5, 84.7, 78.3, 75.9, 72.3, 65.9, 64.8, 55.7, 38.9, 33.6. ³¹P NMR (D₂O): 1.00, 0.25. FAB-HRMS: $[M+1]^+$ calcd for $C_{20}H_{27}N_{10}O_{10}P_2$ 629.1387, found 629.1412. UV (H₂O): λ_{max} 259 (ϵ 24,500).
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